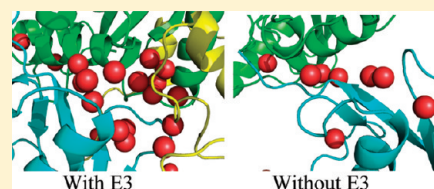


Entropy-Driven Mechanism of an E3 Ligase

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ABSTRACT: Ubiquitin-like modifications are macromolecular chemistry for which our understanding of the enzymatic mechanisms is lacking. Most E3 ligases in ubiquitin-like modifications do not directly participate in chemistry but are thought to confer allosteric effects; however, the nature of the allosteric effects has been elusive. Recent molecular dynamics simulations suggested that an E3 binding enhances the population of the conformational states of the E2·SUMO thioester that favor reactions. In this study, we conducted the first temperature-dependent enzyme kinetic analysis to investigate the role of an E3 on activation entropy and enthalpy. The small ubiquitin-like modifier (SUMO) E3, RanBP2, confers unusually large, favorable activation entropy to lower the activation energy of the reaction. Mutants of RanBP2, designed to alter the flexibilities of the E2·SUMO thioester, showed a direct correlation of their favorable entropic effects with their ability to restrict the conformational flexibility of the E2·SUMO thioester. While the more favorable activation entropy is consistent with the previously suggested role of E3 in conformational selection, the large positive entropy suggests a significant role of solvent in catalysis. Indeed, molecular dynamics simulations in explicit water revealed that the more stable E2·SUMO thioester upon E3 binding results in stabilization of a large number of bound water molecules. Liberating such structured water at the transition state can result in large favorable activation entropy but unfavorable activation enthalpy. The entropy-driven mechanism of the E3 is consistent with the lack of structural conservation among E3s despite their similar functions. This study also illustrates how proteins that bind both SUMO and E2 can function as E3s and how intrinsically unstructured proteins can enhance macromolecular chemistry in addition to their known advantages in protein–protein interactions.



The significant role of water in protein structure and function has been well documented by many elegant studies.^{1–4} The release of ordered water from protein to bulk solvent leads to a small enthalpic penalty due to the loss of hydrogen bonds maintained by ordering of water. However, the entropic gain of releasing ordered water far outweighs the enthalpic loss, resulting in a free energy gain for each sequestered water molecule of as much as 2 kcal/mol.¹ Such an entropy gain contributes greatly to protein folding, binding, and entropy-driven catalysis, such as in peptide bond synthesis on a ribosome.³

Ubiquitin and ubiquitin-like modifications are among the most important post-translational modifications that regulate functions of target proteins.^{5,6} In such a modification process, a ubiquitin-like protein (Ubl) is first activated by an E1 enzyme and then is transferred to E2 by forming a thioester bond with the SH group of the catalytic Cys residue of E2. In the final step, which usually requires the participation of an E3 ligase, the Ubl is attached to target proteins by the formation of an isopeptide bond with a specific Lys residue on the target protein. The E3s that do not form thioester intermediates with Ubl constitute the largest family. These E3s bind to an E2 site that is distal from the catalytic active site^{7,8} and are considered as adaptors for bringing E2 and substrate proteins together. However, many studies have indicated that E3s are not simply adaptors, although the nature of their allosteric effects remains unknown despite extensive structural analysis of E2–E3 complexes and E2·Ubl thioester intermediate mimetics.⁷

One of the outstanding questions in ubiquitin-like modifications is how E3 ligases confer their functions despite their diverse structures ranging from RING domain-containing large proteins or multiprotein complexes to intrinsically unstructured proteins.⁹ While the majority of E3s that do not form thioester intermediates with UbIs contain a RING or U-box E2-binding motif, recent studies have identified a class of SUMO-binding motif (SBM, also known as SUMO-interacting motif, SIM)-dependent E3 ligases that contain a conserved SIM instead of a conserved E2-binding motif. Increasing numbers of SIM-dependent E3 ligases have been identified, including Pc2,^{10–12} a Kaposi's sarcoma-associated herpes virus-encoded protein,¹³ and RanBP2,^{14,15} which is the best characterized member of this class. Both X-ray crystallographic studies and solution NMR studies showed that, like other E3 ligases, RanBP2 binds to a surface of E2 that is distal from the catalytic and substrate-binding sites and thus does not directly participate in the chemistry.^{8,16} RanBP2 offers an attractive model for investigating the allosteric effect of E3 ligases, and understanding its allosteric effect through binding a remote site of E2 has general implications for other E3 ligases.

In this study, we conducted the first temperature-dependent enzyme kinetic analysis of ubiquitin-like modifications to investigate the allosteric effect of RanBP2 on activation entropy and enthalpy. In combination with mutants that were designed to

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alter the conformational flexibility of the SUMO·E2 thioester, measurements of the enthalpic and entropic components of the activation Gibbs free energy revealed that RanBP2 confers an unusual large favorable activation entropy and that RanBP2's allosteric effect is conferred through its ability to alter the conformational flexibility of the E2·SUMO thioester. Consistent with previous molecular dynamic simulations, our simulation also suggested that RanBP2 restricts the flexibility of the E2·SUMO thioester. In addition, we further analyzed structured water and found that the E2·SUMO thioester stabilized by RanBP2 allows sequestration of water molecules in the E2·SUMO thioester conjugates in the ground state. Releasing ordered water to bulk solvent in the transition state due to conformational changes is consistent with the observation of an increased enthalpy cost, but the large favorable entropy gain that lowers the activation energy of the reaction in the presence of the E3. This novel model of E3 function does not require a specific structure of E3 and offers a unifying theme of the similar allosteric effects of E3s in the absence of a conserved structure.

EXPERIMENTAL PROCEDURES

Plasmids. Expression constructs for SUMO E1 (SAE1/SAE2), Ubc9, wild-type and C52A mutant SUMO-1_{1–97}, GST-Sp100_{241–360}, and RanGAP1 were created as previously described.^{16–19} cDNAs encoding the IR1-M domain (residues 2629–2710), the E3_{ΔSIM} (or IR1-M_{ΔSIM}) (residues 2637–2710) domain of RanBP2, and E3_G and E3_{GG} (inserted after residue 2638) followed by His₆ tags at the C-termini were amplified by polymerase chain reaction and inserted between the NdeI and BamHI sites of pET11a+ (Novagen). All constructs were confirmed by DNA sequencing.

Expression and Purification of Recombinant Proteins. Human SUMO-1 (C52A), SUMO E1, Ubc9, and GST-Sp100 were purified using protocols described previously.^{16–18} The various E3 constructs were expressed in *Escherichia coli* BL21-(DE3) cells, purified using a Ni-NTA column (Qiagen), and dialyzed against phosphate-buffered saline (PBS) containing 1 mM DTT. The IR1-M domain of RanBP2 used in NMR analysis was labeled with ²D, ¹⁵N, and ¹³C by growing *E. coli* cells (37 °C) in M-9 minimal medium in D₂O and supplemented with trace minerals and Basal Medium Eagle vitamins (Gibco) using ¹⁵NH₄Cl and [¹³C]glucose as the nitrogen and carbon sources, respectively.

Pull-Down Assay. Protein pull-down assays were performed as previously described.¹⁷ Microtiter plates (Corning Costar) were coated (4 °C, overnight) with the IR1-M or IR1-M_{ΔSIM} domain of RanBP2 (at 0.5 μg/well) in PBS (100 μL/well). Wells were washed twice and blocked (1 h at room temperature) with 150 μL of blocking buffer (2% BSA and 0.1% Tween 20 in PBS). Either 100 ng of purified recombinant Ubc9 or 5 μL of in vitro translated ³⁵S-labeled RanGAP1 sample was added with 100 μL of blocking buffer followed by incubation (1 h at room temperature). Wells were washed three times with 150 μL of blocking buffer and then four times with 150 μL of PBS containing 0.1% Tween 20. Bound Ubc9 and RanGAP1 (or SUMOylated RanGAP1) were extracted with SDS sample buffer and detected by Western blotting using the anti-Ubc9 antibody and autoradiography using a phosphor imager (Molecular Dynamics), respectively.

Kinetic Analysis. The recombinant SUMO-1 C52A mutant was used instead of wild-type SUMO-1 to avoid complications from the SUMO-1–SUMO-1 disulfide-linked dimer. The mutation does not alter conjugation activities.¹⁹ In vitro SUMOylation of Sp100 was conducted via incubating the reaction mixture (4.2 μL) containing 0.78 μM E1, 0.35 μM Ubc9, 22.7 μM SUMO-1(C52A), 3.4 mM ATP, 2.6 μM wild-type and mutant IR1-M domains of RanBP2 (E3, E3_G, E3_{GG}, and E3_{linker}), or assay buffer [5 mM MgCl₂, 0.1% Tween 20, 20 mM Hepes (pH 7.5), and 50 mM NaCl] for non-E3 assays, and SUMOylation was initiated by the addition of GST-Sp100 of various concentrations (2–9 μM) at different temperatures. Prior to the addition of GST-Sp100, mixtures were incubated (10 min at room temperature) for saturation of RanBP2 SUMOylation. A different set of temperatures was used for each E3 condition. For insertion of wild-type E3 and E3_{GG}, the reaction mixtures were incubated at 25, 28, 31, and 37 °C for 3 min. For insertion of E3_{linker} and no E3 (NE) assays, the reaction mixtures were incubated at 31, 37, 40, and 43 °C for 10 min. Reactions were then quenched with 9 μL of 360 mM DTT loading buffer, and components were separated on an SDS–PAGE followed by Western blotting. All experiments were performed in triplicate with the exception of E3_{linker} (in duplicate). The use of different temperatures was due to the vastly different reaction rates of the different E3 variants. The reactions in the presence of E3 and E3_{GG} were very fast; thus, to ensure that product formation is linearly dependent on time over reasonable reaction time for proper estimation of initial rates, they were conducted at lower temperatures. Reactions in the absence of E3 or with E3_{linker} occurred at much slower rates, and thus, higher temperatures and longer reaction time were necessary to obtain detectable product formation.

Western Blot Analysis. Gels were transferred to a PVDF-FL membrane in TOWBIN buffer (containing 192 mM glycine and 25 mM Tris base). The membrane was washed twice with PBS and 0.1% Tween 20 buffer and then incubated with blocking buffer overnight. Free SUMO and SUMOylated products were detected with a mouse anti-SUMO-1 antibody (Abgent, San Diego, CA) and a donkey anti-mouse secondary antibody and imaged and quantitated with an Odyssey scan (Li-Cor Biosciences).

MD Simulations. MD simulations were conducted in explicit water (TIP3P)²⁰ with NAMD.²¹ The simulations were conducted with the Charmm27 force field,²² the isothermal–isobaric (NPT) ensemble, periodic boundaries, and particle mesh Ewald (full) electrostatics calculations.^{23,24} Conjugate gradient minimization was performed for 1000 steps until convergence of force to 0.1 kcal mol^{−1} Å^{−1}. Then, the systems were heated to 310 K in steps of 1 K for every five MD steps. Both systems were equilibrated for 10 ns. The trajectory files were save every 2.5 ps. MD simulations were conducted for 125 ns for the complex without E3 and 150 ns for the complex with E3. MD simulations were prepared by VMD1.8²⁵ and conducted by NAMD2.6 using a local HPC computer cluster. The trajectory analyses were conducted by VMD1.8. Only the last 100 ns simulations were used for the analysis. The thioester reaction site was defined by residues K524 (RanGAP1), C93 (Ubc9), and G97 (SUMO). Water molecules that were within 5.0 Å of these residues and appeared at the same position longer than 10 ns are listed in Table 2. The boundary of the pocket at the SUMO-1–UBC9 interface was defined by residues Q29, D30, S31, I34, Y51, R54, Q55, and R63 of SUMO-1 and K48, K49, T51, A106, T108, K110, Q111, and N121 of UBC9. Water molecules that were within 5.0 Å of these residues and appeared at the same position longer than 10 ns are listed in Table 3.

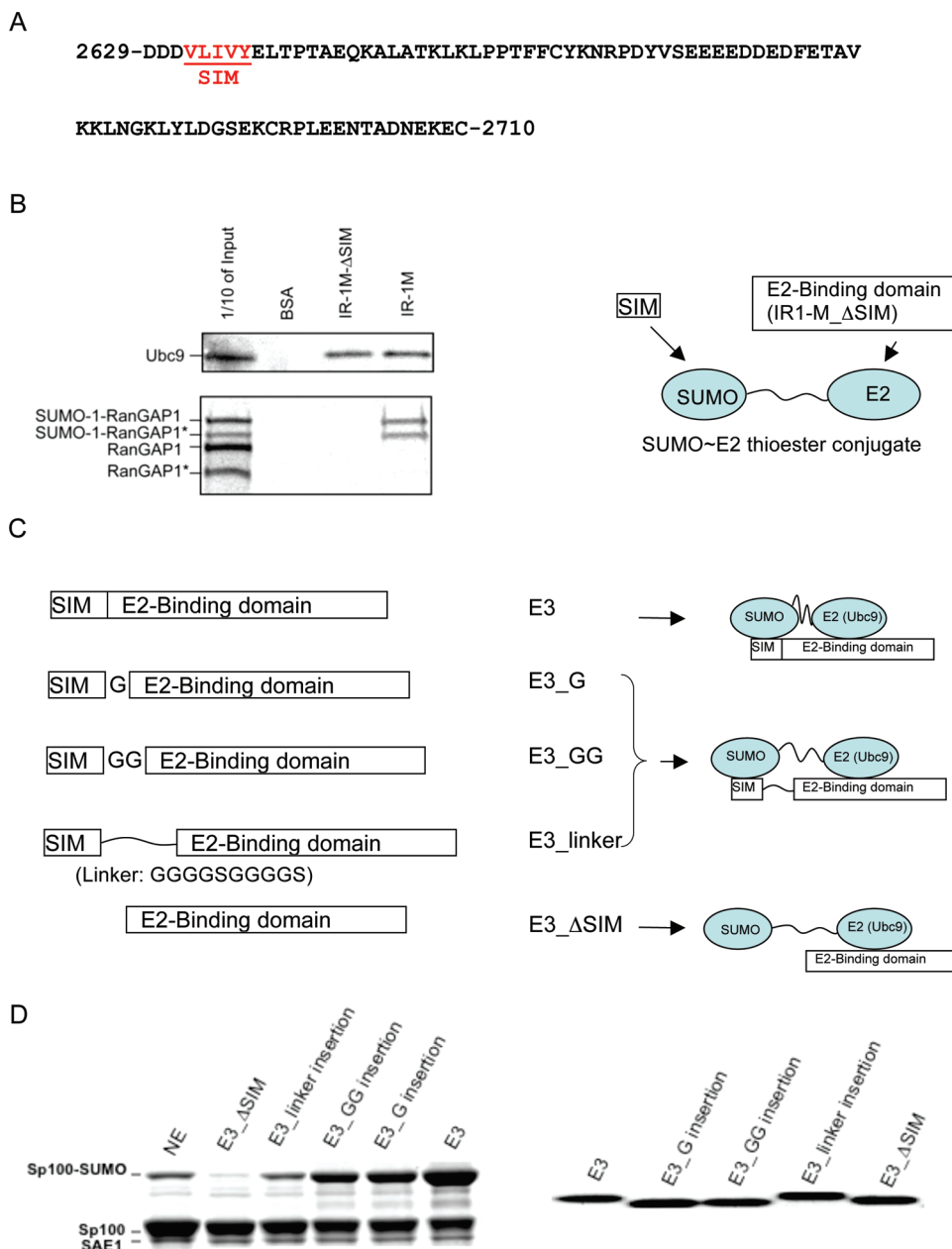


Figure 1. Independence of the SUMO- and Ubc9-binding sites in the IR1-M domain of RanBP2. (A) The IR1-M domain of RanBP2 is rich in Pro, polar, and charged residues, similar to other disordered proteins. The SUMO-binding motif (SIM) is underlined and colored red. (B) The SUMO- and E2-binding regions in E3 are independent of each other. Pull-down experiments demonstrate that the SIM is required for binding SUMO (shown with SUMOylated RanGAP1). However, deletion of SIM does not affect the interaction with Ubc9. Bound Ubc9 was detected by Western blotting. RanGAP1 was labeled with ^{35}S by an in vitro transcription–translation reaction using rabbit reticulocyte extracts in the presence of [^{35}S]methionine and was detected by autoradiography. Both RanGAP1 and SUMOylated RanGAP1 were produced during the translation process because rabbit reticulocyte extracts contained all the necessary components for SUMOylation of RanGAP1.²⁶ Asterisks indicate the truncated version of RanGAP1 due to the presence of an ATG start codon in the middle of RanGAP1 cDNA. (C) Schematic diagrams of the E3 variants that contained linker insertions between the SIM and Ubc9-binding regions. Their expected effects on the SUMO·E2 thioester are shown to the right. (D) Comparison of ligase activity of the different E3 variants. The assay mixture (7 μL) containing 0.78 μM E1, 0.35 μM E2, 2.6 μM E3, 4 mM ATP, 22.7 μM SUMO, and 8.4 μM Sp100 was incubated (37 $^{\circ}\text{C}$ for 14 min) and the reaction quenched with freshly prepared DTT loading buffer (360 mM, 8.5 μL). The panel on the right shows Coomassie staining of the E3 variants to demonstrate that similar amounts were used in the activity assay.

RESULTS

Design of E3 Variants for Examining the Requirements for Ligase Function. Within RanBP2, the IR1-M (internal repeat 1 and M) domain, which contains a SIM and an E2 (Ubc9)-binding segment, confers the E3 ligase activity for SUMOylation

(Figure 1A).¹⁴ IR1-M is highly flexible in solution with the characteristic sequence of a disordered protein but adopts a β -strand upon binding SUMO and forms two α -helices upon binding the E2.^{8,16} Previous structural studies suggested that the SUMO and Ubc9-binding segments of IR1-M do not overlap. To confirm the independence of the SUMO-binding segment

from the E2-binding region, we deleted the SIM from IR1-M; the resulting construct was termed IR1-M_ΔSIM [residues 2637–2710 (Figure 1B)] or E3_ΔSIM (Figure 1C). While pull-down assays showed that the affinity of IR1-M_ΔSIM for Ubc9 was similar to that of IR1-M, IR1-M_ΔSIM failed to bind the SUMO-conjugated protein RanGAP1 (Figure 1B),^{17,26} confirming that the SUMO and E2 binding activities are independent of each other.

IR1-M is a prototype of the SIM-dependent SUMO ligases. To identify factors required for E3 ligase activity in addition to SUMO and Ubc9 binding ability, we created mutations in IR1-M by introducing flexible linkers of various lengths between the E2- and SUMO-binding segments. Three constructs were generated, in which the linkers consisted of one or two Gly residues (G and GG) or 10 residues of the GGGGSGGGGS sequence (Figure 1C). The linker insertion constructs contained the same SUMO- and E2-binding segments. We hypothesized that if E3 were to induce a specific structure of the SUMO·E2 thioester that is required for SUMO transfer, the G and GG mutants were likely to have different orientation effects on the SUMO- and E2-binding segments because of the different number of chemical bonds introduced between the two segments. In addition, variable linker lengths allow the evaluation of how spacing between the SUMO- and Ubc9-binding segments affects E3 ligase activity.

E3_ΔSIM did not display any E3 activity under any conditions (Figure 1D), consistent with the previous finding that both the SUMO- and E2-binding sites are required for E3 ligase activity.^{8,16} However, the longer the linker between SIM- and E2-binding segments, the lower the E3 activity (Figure 1D). Activity levels of E3 variants were in the following order: wild-type IR1-M (E3) > E3_G ≈ E3_GG > E3_linker > E3_ΔSIM ≈ no E3. These results were independent of Sp100 concentration and reaction temperature. Furthermore, because E3_ΔSIM and E3 have the same binding affinity for Ubc9, the fact that E3_ΔSIM showed a complete loss of ligase function indicates that E3's allosteric effect is not on the structure of E2, but on the E2–SUMO thioester. E3_G and E3_GG had similar activities, which correlated with their similar linker length but not their structural effects on SUMO and Ubc9 in the thioester conjugate due to the different number of bonds introduced into the linker. These observations suggest that the allosteric effect of E3 is unlikely to be on the structure of the E2·SUMO thioester. Instead, the E3 activities of the mutants were directly correlated to the linker length: the longer the linker, the lower the activity.

These results suggest a dynamic allosteric effect, the ability of E3 to restrict the flexibility of the E2·SUMO thioester conjugates. Recent NMR studies of various ubiquitin·E2 thioesters revealed that ubiquitin and E2 in their thioester conjugate or its mimetics have significant relative flexibility.^{27–30} The lack of a stable structure of the Ubl·E2 thioester is consistent with a previous molecular dynamic simulation of the E2·SUMO thioester,³¹ although the thioester is not sufficiently stable to be characterized by structural methods.³² The flexibility of the thioester conjugate should be restricted by binding IR1-M. Thus, the linker length-dependent E3 activities suggest that the ligase activity is directly linked to their abilities to restrict the flexibility of the E2·SUMO thioester conjugate.

Entropy-Driven Mechanism of E3. To gain insights into how the flexibility of the E2·SUMO thioester is related to the effect of IR1-M, the entropy and enthalpy components of activation energy of the transfer of SUMO from E2 to a substrate, in the presence and absence of the E3, were measured by temperature-

dependent enzyme kinetic analysis. Such analysis provides the necessary insights into how an E3 lowers the activation energy. Activation enthalpy and entropy can be extracted using the Eyring–Polanyi equation:

$$\ln \frac{k}{T} = -\Delta \frac{H^\ddagger}{R} \times \frac{1}{T} + \ln \frac{k_B}{h} + \Delta \frac{S^\ddagger}{R}$$

where k is the reaction rate constant, R is the gas constant, T is the absolute temperature, h is Planck's constant, and k_B is Boltzmann's constant. Therefore, the enthalpy of activation, ΔH^\ddagger , can be extracted from the slope of the Eyring plots, $\ln(k/T)$ versus $1/T$, and the activation entropy, ΔS^\ddagger , can be extracted from the intercepts.

Activation entropy and enthalpy had not been measured previously for ubiquitin-like modifications, and thus, an innovative approach had to be designed. Although the single-turnover reaction of transferring SUMO from the E2·SUMO thioester to target proteins gives information about the microscopic rate constants,⁸ it is far too quick (20–60 s) to handle the large number of reactions required to assay for multiple temperatures at multiple substrate concentrations, in the absence and presence of wild-type and mutant E3, at the same time in order to eliminate systematic errors. Instead, we chose the measurements of the net transfer rate constants ($k = k_{\text{cat}}/K_m$) for the transfer of SUMO from E2 to substrates using the overall conjugation reactions conducted at limiting E2 concentrations.^{19,33–38} The rate constant also directly reflects the E3-dependent step when the step is rate-limiting of the overall conjugation reactions. In addition, the product formation rate was controlled by the low E2 concentrations, and thus, the initial rates were measured over 3–10 min to eliminate measurement errors arising from insufficient temperature equilibration or slight time variations in sample handling.

The net transfer rate constants ($k = k_{\text{cat}}/K_m$) for the transfer of SUMO from E2 to substrates were measured using the plot of $[E]/V_o$ versus the inverse of substrate concentration (here $1/[\text{GST-Sp100}]$), under conditions in which transfer of SUMO from E2 to substrate is rate-limiting in this multi-enzyme, multi-step modification, as described previously.³⁶ These experiments used Sp100, a well-established SUMO substrate whose SUMOylation is stimulated by RanBP2.^{14,16} Different temperatures and reaction times were chosen for the various E3 variants because they have significantly different reaction rates. Because the E3_G and E3_GG mutants had comparable effects (Figure 1D), E3_GG was chosen to represent the two in temperature-dependent kinetic experiments. In addition, reactions in the absence of E3 are representative of reactions in the presence of E3_ΔSIM, because they have identical effects on SUMOylation rates at various substrate concentrations and reaction temperatures (Figure 1).⁸ For wild-type E3 and E3_GG, the reactions were fast (3 min) and were performed at 25–37 °C. For the E3_linker insertion and no E3 (NE) assays, higher temperatures (31–43 °C) and a longer reaction time (10 min) were used, because the reactions occurred at much slower rates. Conjugation reactions were performed over durations in which product formation was proportional to time and represented initial rates, V_o . Each kinetic experiment involved 64 reactions (four substrate concentrations for four E3 variants at four different temperatures) that were conducted at the same time to minimize systematic errors (representative data shown in Figure 2A). The measurements were repeated three times to obtain

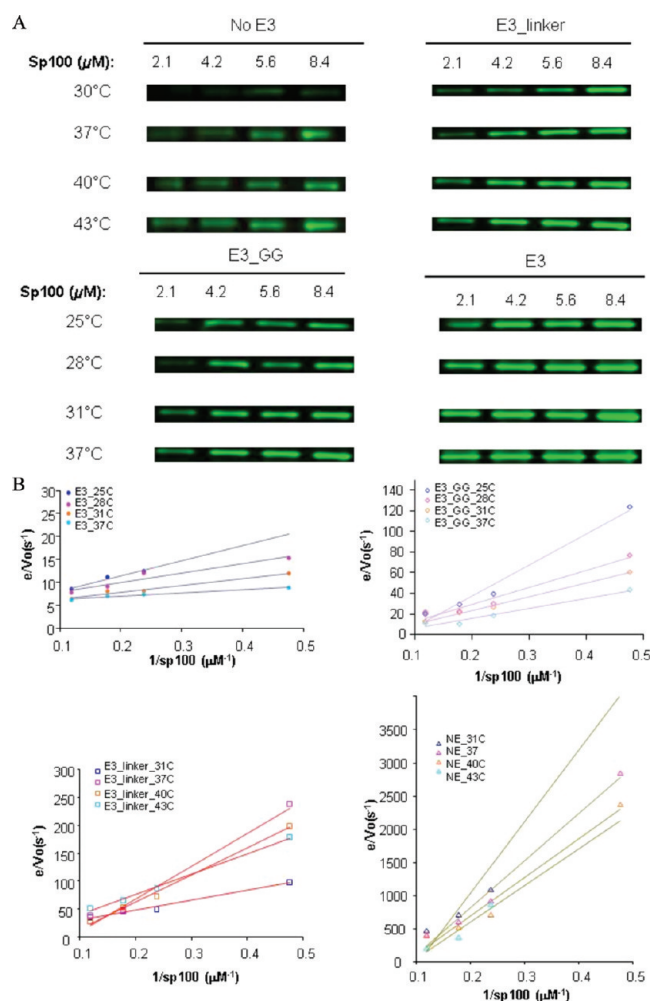


Figure 2. Temperature-dependent enzyme kinetic measurements of wild-type and mutant E3. (A) Examples of Western blots used to detect the Sp100–SUMO product. The differences in temperatures and reaction times between the E3 variants were necessary to ensure that Sp100–SUMO product can be accurately detected. (B) Double-reciprocal plots of data shown in panel A. The assay mixture (4.2 μ L) contained 0.78 μ M E1, 0.35 μ M E2, 2.6 μ M E3, 4 mM ATP, 22.7 μ M SUMO, and Sp100 at four concentrations (2.1, 4.2, 5.6, and 8.4 μ M). The following E3 variants were used: no E3 (NE), E3_linker, E3_GG insertion, and wild-type E3. Temperatures: 31–43 $^{\circ}$ C for 10 min for NE and E3_linker assays and 25–37 $^{\circ}$ C for 3 min for E3_GG insertion and wild-type E3 assays. Assays were quenched with freshly prepared 360 mM DTT loading buffer (8.5 μ L), separated by SDS–PAGE, and detected by Western blotting using an anti-SUMO-1 antibody.

uncertainty of the measurements. A couple of points (of 64) were dropped in each repeat because of band intensities at low substrate concentrations that were too low to measure accurately. Plots of $[E]/V_0$ versus $1/[GST\text{-}Sp100]$ were linear, indicating that the conditions were appropriate for Michaelis–Menten kinetics (Figure 2B).

The double-reciprocal plots fit well to a line (Figure 2B) based on R values, and the slopes of the double-reciprocal plots, corresponding to K_m/k_{cat} ratios, were well determined. The K_m/k_{cat} ratios correspond to the net transfer rate constant for the transfer of SUMO from E2 to substrates. We found that the k_{cat}/K_m values were $(1.7 \pm 0.6) \times 10^2$ and $(1.1 \pm 0.4) \times 10^5$ $M^{-1} s^{-1}$ in the absence and presence of IRI-M, respectively

Table 1. Summary of Measured Rates, Activation Entropies, and Activation Enthalpies of the Reactions and Calculated Activation Free Energies in the Absence of E3 and in the Presence of Wild-Type E3 and Different E3 Mutants with Insertions

	k ($M^{-1} s^{-1}$) ^a	ΔH^{\ddagger} (kcal/mol)	$T\Delta S^{\ddagger}$ (kcal/mol) ^a	ΔG^{\ddagger} (kcal) ^a
E3	$(1.1 \pm 0.4) \times 10^5$	24.0 ± 2.4	12.9 ± 2.2	11.0 ± 3.3
E3_GG	$(1.2 \pm 0.5) \times 10^4$	14.7 ± 2.7	2.2 ± 2.4	12.4 ± 3.6
insertion				
E3_linker	$(1.8 \pm 0.6) \times 10^3$	12.0 ± 1.1	-1.4 ± 0.9	13.6 ± 1.4
insertion				
no E3 activity	$(1.7 \pm 0.6) \times 10^2$	10.7 ± 0.3	-4.3 ± 0.3	15.0 ± 0.4

^a Values correspond to those measured at 37 $^{\circ}$ C.

(Table 1). These results are very consistent with published values that were independently determined using single-turnover reactions (transferring SUMO from the E2·SUMO thioester to substrate). Recent measurements of k_2/K_d , which is similar to k_{cat}/K_m in the absence and presence of IRI-M were $(2.3 \pm 0.3) \times 10^2$ and $(9.4 \pm 0.9) \times 10^4$ $M^{-1} s^{-1}$, respectively, for a substrate (MEF2p) that has a modification efficiency similar to that of Sp100.³⁹ Thus, our results have been independently validated.

Eyring analysis was conducted to extract the activation entropy and enthalpy for reactions with and without E3 (Figure 3 and Table 1). The activation entropy, $T\Delta S^{\ddagger}$, without E3 showed a negative value (-4.3 kcal/mol), which suggests that the transition state is more orderly than the ground state of reactants in the absence of E3. Interestingly, reactions in the presence of wild-type E3 showed a large favorable activation entropy $T\Delta S^{\ddagger}$ (12.9 kcal/mol) that is the driving force for the lowered activation Gibbs free energy to accelerate the reaction. The entropic advantage achieved by the E3 was reduced when flexible linkers were inserted between the SUMO- and E2-binding sites: the longer the linker, the smaller the entropic advantage achieved by the E3 (Table 1). In contrast, reactions conducted with E3 had higher (unfavorable) activation enthalpies (ΔH^{\ddagger}) than reactions without E3. ΔH^{\ddagger} was lower when the E3 contained linker insertions, and the longer the flexible linker, the lower the ΔH^{\ddagger} , with ΔH^{\ddagger} being the lowest for reactions without E3 or with E3_ΔSIM (Table 1).

Effect of E3 on the Stability of the SUMO·E2 Thioester and Its Interaction with Water. To gain insight into the origin of the unfavorable change in the activation enthalpy and the favorable change in the activation entropy, we conducted molecular dynamics (MD) simulations. It was not feasible to investigate the SUMO·Ubc9 thioester conjugate experimentally, because it converts to a Ubc9–SUMO isopeptide bond within 1 h.³² In addition, thioester mimics using either an ester bond or a disulfide bond had low solubility, likely due to the noncovalent interaction between SUMO and the N-terminal region of Ubc9,⁴⁰ resulting in the formation of large aggregates of the thioester mimic at concentrations required for structural studies.

As described in a recent report, the structure of RanBP2 in complex with SUMOylated RanGAP1 and Ubc9 was used to deduce the structure of the E2·SUMO thioester in complex with the E3.^{8,31} Two MD simulations were designed to mimic the thioester conjugate: one with the E3 and the other with the E3 deleted. To mimic the SUMO·Ubc9 thioester, the isopeptide

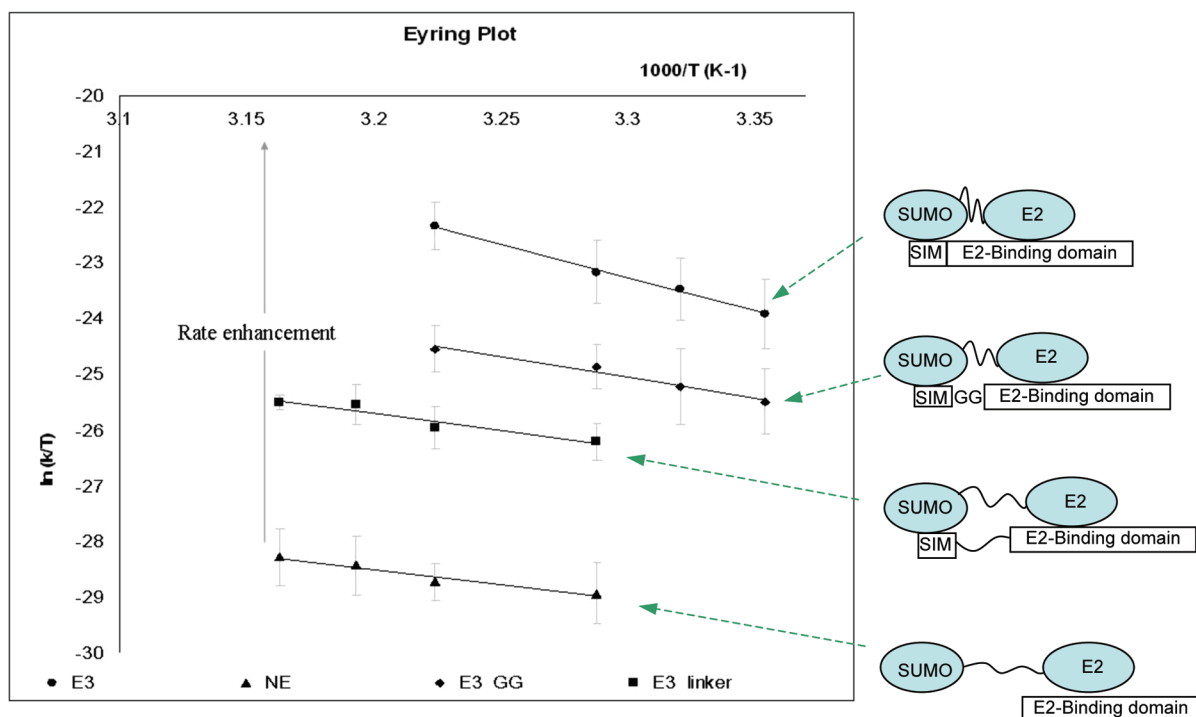


Figure 3. Eyring plots of kinetic experiments. Slopes of the plot were used to calculate the activation enthalpy, and the intercepts were used to calculate the activation entropy. Included are schematic drawings of each E3 variant and their expected effects on the E2·SUMO thioester conjugate.

bond between SUMO-1 (G97) and RanGAP1 (K524) was broken, and a thioester bond between SUMO-1 and Ubc9 was constructed without significant structural alteration, because the two atoms to be bonded are already in the proximity of each other (~ 3.5 Å). The simulation was conducted for 125 ns for the complex without E3 and for 150 ns for the complex with E3.

As found in the recent report, analysis of the structures in MD simulations confirmed that the SUMO-1·Ubc9 thioester explores a much smaller conformational space in the presence of E3 than in the absence of E3 (Figure 4A,B). The relative flexibility of SUMO and the E2 in their thioester conjugate is consistent with experimental observations of ubiquitin·E2 thioester conjugates^{27–30} and is expected due to the lack of extensive contacts between SUMO and Ubc9 that are needed for a stable structure. The flexibility observed for the E2·SUMO thioester conjugate in the MD simulation is not as extensive as that observed by NMR and small-angle X-ray scattering studies of ubiquitin·E2 thioester mimetics,²⁷ likely because of the short simulation time. The more favorable activation entropy in the presence of the E3 than in the absence of the E3 is consistent with the recently reported MD studies suggesting that E3 binding may select the population that favors the reaction and substrate binding. In addition, if the transition state remains the same in the absence and presence of E3, the more ordered thioester reactive center in the ground state in the presence of E3 than in the absence of E3 would reduce the entropic penalty for the formation of the transition state. However, this model does not fully account for the large positive activation entropy observed above.

The large positive activation entropy, which indicates that the ground state is more orderly than the transition state, suggests ordered water is released upon formation of the transition state.

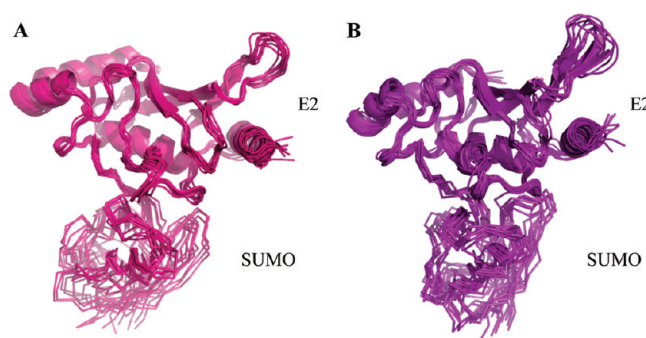


Figure 4. Superposition of the structures of the E2·SUMO thioester conjugate from the last 50 ns of simulations in the absence (A) and presence (B) of the E3. A structure from every 5 ns was taken for the superposition. The superposition minimized the root-mean-square deviation of the E2 only; thus, the fluctuation in the SUMO structures reflects variations in the relative positioning of the two proteins in the thioester conjugate.

Thus, we conducted a detailed analysis of the residence time of water molecules and locations of structured water along the MD trajectory. The more stable E2·SUMO thioester in the presence of the E3 results in more stably bound water at the thioester reaction site as well as at the SUMO–E2 interface. Table 2 shows the residence time of water molecules that stayed within 5 Å of residues involved in the chemistry, K524 (RanGap1), C93 (Ubc9), and G97 (SUMO-1), for longer than 10 ns during the simulation. The E3-bound structure not only has more water molecules that stayed for more than 10 ns near the thioester bond, but also with longer residence time. The waters are found to form hydrogen bonds with the polar side chains and backbone. Snapshots of the MD trajectory illustrate the positions of some of

these bound waters in the vicinity of the thioester bond in the presence of the E3 (Figure 5A) and in the absence of the E3 (Figure 5B). The long-residence water molecules did not always stay in a particular hydrogen bonding pattern but stayed in the same area. Conformational changes upon the formation of the transition state likely cause the release of water bound at

Table 2. List of Water Molecules within 5 Å of Residues Involved in the Chemistry, K524 (RanGAP1), C93 (UBC9), and G97 (SUMO-1), and Their Residence Times^a

without E3		with E3	
Atm_index	time (ns)	Atm_index	time (ns)
		44573	27.9
		30239	24.5
16740	17.7	35090	42.8
26157	15.8	62360	35.9
34425	13.8	43952	17.7
39090	13.0	46241	13.4
17643	11.1	47168	12.2
27312	11.0	15878	11.6

^a The water molecules in the top box are specifically located at the interface between RanGAP1 and UBC9.

the ground state. A similar mechanism has been implicated for other enzyme catalytic mechanisms such as ribosomal catalysis of peptide synthesis.^{41–45} Two structured water molecules were also found at the interface between RanGAP1 and Ubc9 (E2), which may contribute to the slightly enhanced affinity (approximately 7-fold) of the substrate for E2 in E3-stimulated reactions.⁸ These water molecules were not present in the crystal structure of SUMOylated RanGAP1 in complex with Ubc9 and RanBP2, possibly because of the lack of the thioester bond and experimental conditions.

E3 may also contribute to an overall entropic force of the reaction through water molecules that are stabilized at the interface between SUMO-1 and Ubc9 (Table 3 and Figure 6). Along the MD trajectory, more than twice as many water molecules (as many as 44) stayed in a “water pocket” at their interfaces for more than 10 ns in the presence of the E3 than in the absence of the E3, apparently because of the more stable structure of the thioester conjugate in the complex with the E3. Several water molecules in this location were observed in the crystal structure of SUMOylated RanGAP1 in complex with Ubc9 and RanBP2.⁸ These “trapped” water molecules are expected to be lost upon the completion of the reaction when SUMO-1 dissociates from E2 and contribute to the overall entropic force of the E3-stimulated reaction.

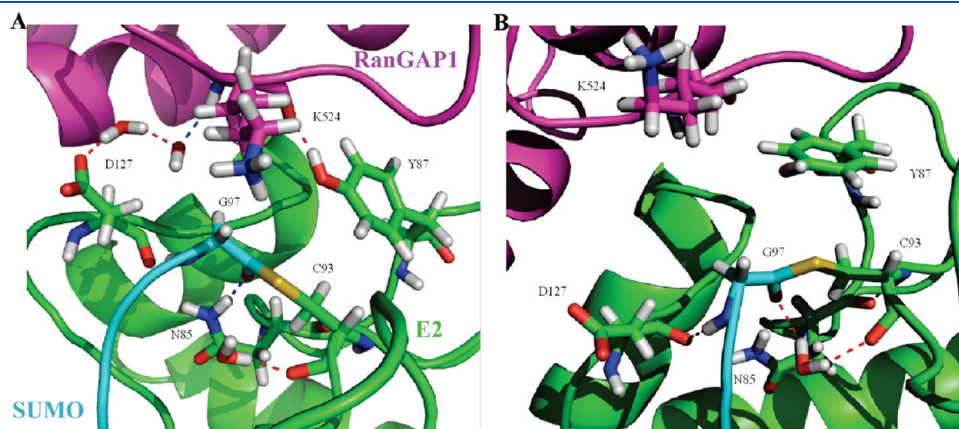


Figure 5. Molecular dynamics simulations for obtaining insights into the entropy-driven mechanism. Snapshots of the molecular dynamics trajectory to show the water molecules in the vicinity of the thioester bond that stayed for more than 10 ns in the presence (A) and absence (B) of the E3. Side chains of key residues are shown: N85, Y87, C93, and D127 of the E2 (green), which are important in catalysis, conjugation residue G97 of SUMO (cyan), and modification residue K524 of RanGAP1 (magenta).

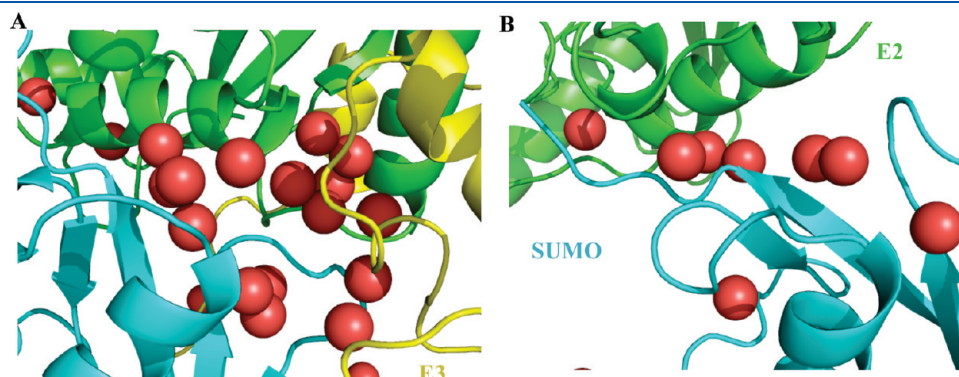


Figure 6. Snapshots of the molecular dynamics trajectory showing the water molecules at the interface between SUMO and Ubc9 in the presence (A) and absence (B) of the E3. Water molecules are shown as red spheres, and SUMO, Ubc9, and the E3 are shown with the same color codes as their labels.

Table 3. List of Water Molecules That Are Trapped in a Pocket at the Interface of UBC9 and SUMO-1 for More Than 10 ns and Their Residence Times^a

without E3		with E3	
Atm_index	time (ns)	Atm_index	time (ns)
17007	46.1	21407	81.5
48039	31	38807	81.4
35412	28.5	17408	56.9
24117	19.4	40091	46.7
28446	17.4	50282	45.2
38376	17.4	41810	35.3
19854	17.1	35387	28.3
44481	15.7	32966	23.9
36753	14.3	60995	22.7
6411	13.3	29270	22.4
13491	13.3	50009	21.6
45138	12.8	42896	21.4
23046	11.7	51074	17.8
29730	11.5	8669	17.4
54999	11.2	54833	16.3
18270	11.1	32972	15.1
15273	11	50132	14.3
52158	10.7	26162	13.9
8346	10.4	24584	13.4
49590	10.1	39071	13
		26519	12.8
		63854	12.5
		48206	12.4
		48239	12.1
		25697	11.8
		61010	11.5
		32090	11.2
		21989	11.2
		22433	11.2
		53285	11.2
		33860	11.1
		26720	11
		11285	11
		50135	11
		12053	10.9
		14066	10.8
		38627	10.7
		25148	10.6
		37670	10.5
		43919	10.5
		28493	10.5
		29126	10.1
		30086	10.1
		57053	10

^a The definition of the SUMO–Ubc9 interface is given in Experimental Procedures.

DISCUSSION

In this study, we have measured the activation entropy and enthalpy for ubiquitin-like modifications for the first time and

found that the E3, RanBP2, confers a large entropic effect to lower the activation energy, thereby accelerating the reactions. Molecular dynamic simulations suggest that solvent effect likely makes a major contribution to the entropy-driven mechanism. The more stable structure of the E2·SUMO thioester conjugate in the presence of E3 than in the absence of E3 traps more stably bound water. Liberating such structured water to bulk water at the transition state can reduce order and result in a favorable activation entropy.¹ Such model is also consistent with the observation of a less favorable activation enthalpy in the presence of E3 than in the absence of E3, which may reflect the breakage of hydrogen bonds of ordered water molecules. This study does not exclude the contributions of other factors to the favorable activation entropy but suggests a novel concept in E3 catalysis that is consistent with all experimental and simulation studies.

The effect of the E3 on activation entropy and enthalpy suggests that the E3 unlikely confers a structural allosteric effect to improve the interaction of the E2 with the transition state, like many well-studied enzymes, because the activation enthalpy, which reflects direct interactions through hydrogen bonding, hydrophobic, and ionic interactions, is more unfavorable in the presence than in the absence of the E3. In addition, the similar activity of E3_G and E3_GG suggests that the determinants of the E3 ligase activity do not lie in its ability to have a structural effect on the E2·SUMO thioester. Instead, the linker length-dependent activities of the E3 variants are consistent with a dynamic effect. Taken together, these data suggest that the ability of RanBP2 to activate SUMOylation depends on its ability to restrict the flexibility of E2·SUMO thioester conjugates.

The novel concept proposed here offers a unifying theme for the allosteric effect of E3. Enthalpic effects are generally evident from X-ray and NMR structures, which include hydrogen bonds and hydrophobic and charge interactions. Entropic effects, on the other hand, arise from a diverse combination of factors due to the solvent and energetics of conformational ensembles, as discussed above. Entropic effects explain the poor correlation of structure and affinity of E2–E3 complexes with the activities of E3. Previous studies have also indicated that the E2-bound ubiquitin plays an important role in E3-stimulated ubiquitination reactions.^{46,47} Even polycations or Zn²⁺ ion can stimulate ubiquitination reactions,^{48,49} and they can reduce the flexibility of the E2·Ubl thioester by charge interactions. The E3 ligase activity of many ubiquitin E3 ligases depends on their affinity for both E2 and target proteins.⁶ When the reactants (the ubiquitin·E2 thioester and a target protein) are brought together, there is a significantly reduced loss of rotational and translational entropies of the substrates upon formation of the transition state, contributing to a favorable activation entropy. Overall, the entropy-driven mechanism, particularly the solvent effect due to the stabilized E2·SUMO thioester by E3 binding, does not require a conserved structure of E3 and may be a general theme of E3's allosteric effect.

This study also provides further mechanistic insights into the action of E3 ligases that lack a conserved E2-binding motif but contain a SIM, suggesting that a protein containing a SIM can be a potential E3 ligase if it also binds Ubc9. Although only a handful of proteins have been identified as SIM-dependent E3 ligases, this study supports the possibility that many more such E3 ligases exist.^{11–13,50}

This study also draws attention to a previously unappreciated function of intrinsically disordered proteins (IDPs), activating macromolecular chemistry, such as ubiquitin-like modifications.

The known prevailing function of IDPs is to bind multiple proteins simultaneously.⁵¹ Thus, they frequently mediate signal transduction pathways and function as scaffolds for protein complex formation during transcription and DNA repair. The ability of IR1-M, an IDP, to bind two proteins, SUMO and Ubc9, is required for its E3 ligase activity. This study has shown how IDPs, by binding multiple proteins, can activate macromolecular chemistry instead of merely functioning as scaffold proteins in these events. We hope that this work stimulates interest in conducting further theoretical and experimental studies to investigate the generality of the entropy-driven mechanism of E3 ligases in ubiquitin-like modifications.

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ABBREVIATIONS

SUMO, small ubiquitin-like modifier; SIM, SUMO-interacting motif, which is also known as SUMO-binding motif (SBM); E1, activating enzyme for ubiquitin-like post-translational modifications; E2, conjugation enzyme for ubiquitin-like modifications; E3, ligase for ubiquitin-like modifications; SAE, SUMO-activating enzyme; RanBP2, Ran binding protein 2; GST, glutathione S-transferase; IR1, internal repeat domain 1 of RanBP2; PBS, phosphate-buffered saline; DTT, dithiothreitol; NMR, nuclear magnetic resonance; SDS-PAGE, sodium dodecyl sulfate-polyacrylamide gel electrophoresis.

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